# SURFACE WATER-QUALITY CHARACTERISTICS IN THE UPPER NORTH FORK GUNNISON RIVER BASIN, COLORADO By J. Michael Norris

# U.S. GEOLOGICAL SURVEY

Water-Resources Investigations Report 86-4152



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# CONTENTS

	Page
Abstract	
Introduction	_
Description of the upper North Fork Gunnison River basin study area	
Data collection	
Water-quality sampling sites	5
Water-quality data collected	5
Water-quality characteristics	5
Dissolved solids and specific conductance	9
Major dissolved constituents	12
Suspended sediment and trace elements	21
Effects of Paonia Reservoir	30
Effects of coal mining	30
Hubbard Creek	
North Fork Gunnison River	37
Summary and conclusions	41
References	42
FIGURES	
	Page
Figures 1-3. Maps showing:	rage
1. Location of upper North Fork Gunnison River	
study area	3
2. Location of water-quality sampling sites	
3. Location of subareas	. 8
4-12. Graphs showing:	J
4. Relation between streamflow and dissolved-solids	
concentration by subarea	10
5. Relation between specific conductance and	
dissolved solids by subarea	13
6. Relation between specific conductance and	
calcium concentration by subarea	16
7. Relation between specific conductance and	10
magnesium concentration by subarea	19
8. Relation between specific conductance and	
sodium concentration by subarea	20
9. Relation between specific conductance and	
alkalinity by subarea	23
10. Relation between streamflow and dissolved-solids	
concentration for upper and lower Hubbard Creek	
near Bowie	37
11. Frequency of occurrence of sulfate concentrations	3,
for upper and lower Hubbard Creek, upstream	
and downstream from coal-mining activities	38
12. Frequency of occurrence of sulfate concentrations	33
for sites on the North Fork Gunnison River,	
upstream and downstream from coal-mining	
activities	40
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#### TABLES

		Page
Table 1.	Water-quality sampling sites	6
2.	Water-quality constituents sampled	7
3.	Regression equations to predict dissolved-solids concentration from streamflow	9
4.	Dissolved-solids concentration statistics by subarea	_
5.	Specific-conductance statistics by subarea	
	Coloium compact matrice statistics by subarea	
6.	Calcium concentration statistics by subarea	15
7.	Magnesium concentration statistics by subarea	17
8.	Sodium concentration statistics by subarea	18
9.	Alkalinity concentration statistics by subarea	22
10.	Potassium concentration statistics by subarea	24
11.	Chloride concentration statistics by subarea	25
12.	Silica concentration statistics by subarea	26
13.	Sulfate concentration statistics by subarea	27
14.	Suspended-sediment concentration statistics by subarea	28
15.	Total-iron concentration statistics by subarea	29
16.	Total-manganese concentration statistics by subarea	31
17.	Total-zinc concentration statistics by subarea	32
18.	Total-lead concentration statistics by subarea	33
19.	Total-cadmium concentration statistics by subarea	34
20	Effects of Paonia Reservoir on water quality of Muddy Creek	35

## METRIC CONVERSION FACTORS

$ exttt{Multiply}$	Вy	To obtain
acre-foot (acre-ft)	0.001233	cubic hectometer
<pre>cubic foot per second   (ft³/s)</pre>	0.02832	cubic meter per second (m <sup>3</sup> /s)
foot (ft)	0.3048	meter
inch per year (in/yr)	2.54	centimeter per year
mile (mi)	1.69	kilometer (km)
square mile (mi²)	2.590	square kilometer (km²)
ton (short)	907.2	kilogram
ton per day (ton/d)	907.2	kilogram per day

Degree Fahrenheit (°F) may be converted to degree Celsius (°C) by using the following equation:

$$^{\circ}C = 5/9 \ (^{\circ}F-32)$$

The following terms and abbreviations also are used in this report:

milligrams per liter (mg/L) micrograms per liter ( $\mu$ g/L) microsiemens per centimeter ( $\mu$ S/cm)

<u>Sea level</u>: In this report "sea level" refers to the National Geodetic Vertical Datum of 1929 (NGVD of 1929)--a geodetic datum derived from a general adjustment of the first-order level nets of both the United States and Canada, formerly called "Mean Sea Level of 1929."

#### SURFACE WATER-QUALITY CHARACTERISTICS IN THE UPPER

NORTH FORK GUNNISON RIVER BASIN, COLORADO

By J. Michael Norris

#### **ABSTRACT**

Analyses of water-quality data collected during 1982 and 1983 in the upper North Fork Gunnison River basin indicate that dissolved-solids concentrations are relatively small, with a mean value near 97 milligrams per liter. Most major dissolved constituents also had small measured concentrations throughout the study area. Trace-element concentrations generally were small; however, total-iron concentration generally was large in the area with a mean concentration of about 8,250 micrograms per liter. The cause of this larger iron concentration probably is related to the local geology.

Paonia Reservoir, located on Muddy Creek, greatly reduced suspended-sediment and trace-element concentrations. The reservoir had only a slight effect on major dissolved-constituent concentrations.

Analyses of alkalinity, sulfate, and dissolved-solids concentrations indicated that little, if any, changes in water quality occur as a result of coal mining; however, more data are needed to make more definite conclusions. Sulfate concentrations increased slightly downstream through the mined area; however, with the small concentrations measured and limited quantity of data, the source of the increased sulfate could not be determined.

## INTRODUCTION

Mining of coal in the upper North Fork Gunnison River basin started in 1903 and continues to the present (1984). To determine the effect coal mining might have on water resources, the background, or natural, water-quality conditions must be known. However, in the upper North Fork Gunnison River basin, as in many other coal-mining areas, little hydrologic data are available to describe either natural conditions or any effects from man's activities.

The objective of this report is to describe the water-quality characteristics of the North Fork Gunnison River basin upstream from Paonia. This report also presents water-quality information from specific sampling sites, upstream and downstream from coal mining, to evaluate the effect coal mining might have on the area's water quality.

#### DESCRIPTION OF THE UPPER NORTH FORK GUNNISON RIVER BASIN STUDY AREA

The study area is located in western Colorado in parts of Delta and Gunnison Counties that include the North Fork of the Gunnison River drainage basin upstream from Paonia (fig. 1). Generally, the 653 mi<sup>2</sup> drainage area is characterized by steep slopes. However, rolling hills are found at lower elevations. The elevation ranges from 13,058 on Mount Owens to 5,608 ft on the North Fork of the Gunnison River in the study area outlet near Paonia.

Climate and vegetation are strongly influenced by elevation and vary throughout the study area. Climate is semiarid at lower elevations and alpine at higher elevations. Precipitation averages about 25 in/yr over the basin, with most precipitation, especially at higher elevations, in the form of snow. Air temperature in the area varies from below zero in the winter to over 90 °F in the summer.

Vegetation at lower elevations consists of grasses and shrubs, with deciduous trees along stream channels. Major vegetation at higher elevations are evergreen forests, with the dominant trees being pine, spruce, fir, and aspen.

Major geologic formations in the area are the Tertiary Wasatch Formation and underlying Cretaceous Mesaverde Formation, including the Ohio Creek Member at the top. These sedimentary formations contain claystone, siltstone, sandstone, conglomerate, and shale. Other geologic units in the area include the Cretaceous Mancos Shale, underlying the Mesaverde, middle Tertiary intrusive rocks, and numerous landslide deposits (Tweto and others, 1976, 1978).

Underground coal mining started in the basin in 1903 and by 1982 production totaled 39,850,000 short tons, with 2,649,400 short tons mined in 1982 (Rushworth and others, 1984). Coal in the study area is in the Somerset coal field and is bituminous. Rushworth and others (1984) estimated that coal reserves in 1982 were 3,348,000,000 short tons.

Paonia Reservoir is the major hydrologic feature in the study area. The dam was built in 1962 by the U.S. Bureau of Reclamation. Paonia Reservoir dam is located on Muddy Creek approximately one-half mile upstream from the confluence of Muddy Creek and Anthracite Creek, where they form the North Fork Gunnison River. The dam has a structural height of 177 ft. The reservoir has a maximum capacity of 23,230 acre-ft, and a normal capacity of 20,950 acre-ft. The major uses of the reservoir's water are irrigation, recreation, water supply, and flood control.

#### DATA COLLECTION

Four continuous streamflow-gaging stations currently (1984) are in operation in the study area and are shown in figure 2. One of the gaging stations is a U.S. Geological Survey station and the other three stations are operated by the State of Colorado. The U.S. Geological Survey station, North Fork Gunnison River near Somerset (fig. 2, site 4) has streamflow records from

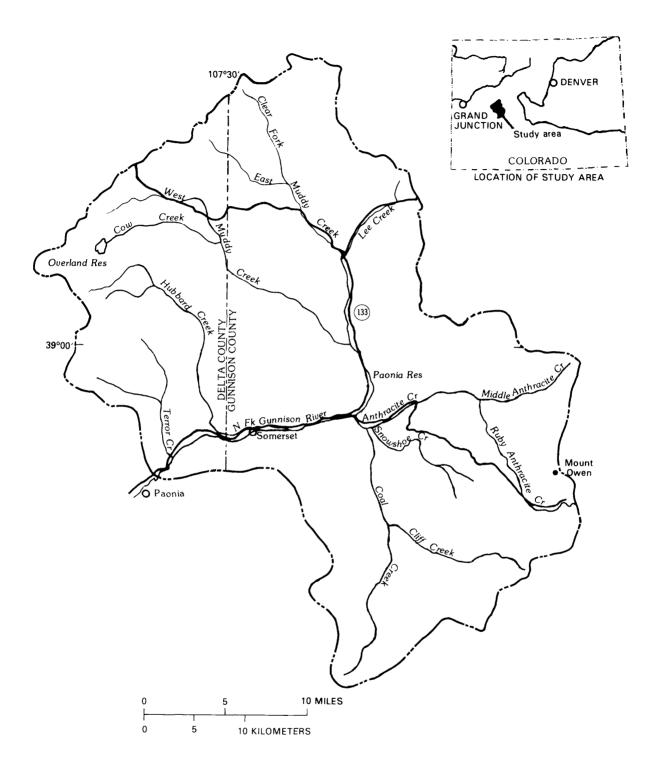


Figure 1.--Location of upper North Fork Gunnison River study area.

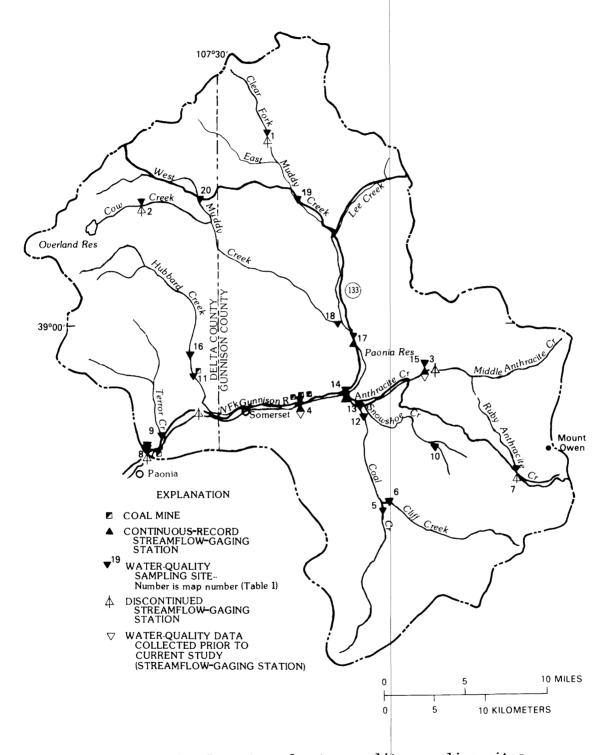


Figure 2.--Location of water-quality sampling sites.

1934 to the present (1984), and water-quality data have been collected periodically. The three State-operated streamflow-gaging stations, upper Anthracite Creek near Somerset (fig. 2, site 15), Muddy Creek below Paonia Reservoir (fig. 2, site 14), and Muddy Creek above Paonia Reservoir (fig. 2, site 17), only have streamflow data. However, a discontinued U.S. Geological Survey streamflow-gaging station, Anthracite Creek near Somerset (fig. 2, site 3) has streamflow and water-quality data from 1977 to 1981, at a point approximately one-half mile upstream from the State-operated Anthracite Creek streamflow-gaging station. Other U.S. Geological Survey discontinued streamflow-gaging stations in the area include Clear Fork near Ragged Mountain (fig. 2, site 1), that operated from 1965 to 1973, Cow Creek near Paonia (fig. 2, site 2), that operated from 1969 to 1982, and Ruby Anthracite Creek near Kebler Pass (fig. 2, site 7).

## Water-Quality Sampling Sites

Nineteen water-quality sampling sites were established in the area for this study (fig. 2). The location and drainage area are listed in table 1. One site shown in figure 2 and in table 1, Anthracite Creek near Somerset (fig. 2, site 3), was not used as a water-quality sampling site for this study, but the data from this site are included in the analyses in this report because both streamflow and water-quality data are available from 1977 to 1981.

The water-quality sampling sites were selected to account for variations in geology, vegetation, elevation, and land use. Because of high elevations in parts of the area, accessibility also was considered in site location. In early spring some areas at higher elevations are inaccessible because of snow depth.

# Water-Quality Data Collected

Water-quality data were collected synoptically starting in the spring of 1982 through the fall of 1983. Water-quality constituents measured are listed in table 2. All major constituents in this report were analyzed for the dissolved concentrations and the trace elements were analyzed for the total concentrations. Samples were collected throughout the range of flows to determine the variability of the water quality during different seasons and for different flow conditions. Samples for water-quality analyses were collected by standard procedures (Brown and others, 1970), and analyzed by the U.S. Geological Survey Laboratory in Denver, using established methods (Skougstad and others, 1979). Measurements of streamflow, pH, water temperature, and specific conductance were made at the time of sample collection. Data collected during this study are presented in Norris and Maura (1985).

### WATER-QUALITY CHARACTERISTICS

For water-quality data interpretation, the study area was divided into four subareas (fig. 3), based on differences in geology and elevation. In the following discussion, the geologic formations for the subareas are listed by relative occurrence within each subarea. Geology of the Muddy Creek subarea

Table 1.--Water-quality sampling sites
[mi², square miles]

Drain- age area (mi <sup>2</sup> )	38.5 12.0 94.6 526 52.3	35.1 20.7 653 29.5 2.35	55.2 101 130 257 95.0	52.0 257 98.4 81.6 27.7
Longi- tude	107°25'50" 107°35'02" 107°16'23" 107°26'53" 107°19'03"	107°19'03" 107°09'47" 107°34'51" 107°33'40" 107°16'14"	107°31'04" 107°20'19" 107°20'24" 107°21'20" 107°16'26"	107°31'51" 107°21'08" 107°21'27" 107°24'19" 107°31'25"
Lati- tude	39°08'36" 39°06'15" 38°57'14" 38°55'45" 38°50'33"	38°50'37" 38°51'46" 38°53'08" 38°54'14" 38°55'06"	38°55'32" 38°55'34" 38°55'38" 38°56'26" 38°57'12"	38°57'41" 38°59'03" 39°00'00" 39°06'20" 39°06'58"
Station name	Clear Fork near Ragged Mountain Cow Creek near Paonia Anthracite Creek near Somerset North Fork Gunnison River near Somerset Upper Coal Creek near Somerset	Cliff Creek near Somerset Ruby Anthracite Creek near Kebler Pass North Fork Gunnison River above Paonia Terror Creek near Paonia Grouse Spring Creek near Marcelling Mountain	Lower Hubbard Creek near Bowie Lower Coal Creek near Somerset Lower Anthracite Creek near Somerset Muddy Creek below Paonia Reservoir Upper Anthracite Creek near Somerset	Upper Hubbard Creek near Bowie Muddy Creek above Paonia Reservoir Lower West Muddy Creek near Paonia Reservoir East Muddy Creek near Ragged Mountain West Muddy Creek near West Muddy Creek Ranger Station
Station r number	09129800 09131100 09132050 09132500 385033107190300	385037107190300 385146107094700 385308107345100 385414107334000 385506107161400	385532107310400 385534107201900 385538107202400 385626107212000 385712107162600	385741107315100 385903107210800 39000107212700 390620107241900 390658107312500
Site	1 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7	6 7 8 9 10	11 12 13 14 15	16 17 18 19 20

Table 2.--Water-quality constituents sampled1

Major constituent	Trace elements
(dissolved concentration)	(total concentration)
Silica	Aluminum
$Nitrogen (NO_2+NO_3)$	Arsenic
Chloride	Cadmium
Fluoride	Chromium
Sulfate	Cobalt
Calcium	Copper
Magnesium	Iron
Potassium	Lead
Sodium	Manganese
Alkalinity (CaCO <sub>3</sub> )	Mercury
	Selenium
	Zinc

<sup>&</sup>lt;sup>1</sup>pH, water-temperature, specific-conductance, and suspended-sediment data were also collected.

consists mainly of the Wasatch Formation and Ohio Creek Member at the top of the Mesaverde Formation, with some young glacial drift and landslide deposits. Geology of the Anthracite Creek subarea is more complex, consisting of young glacial drift, the Wasatch Formation, Mesaverde Group undivided, or Mesaverde Formation, including the Ohio Creek Member at the top, some Mancos Shale, and middle Tertiary intrusive rocks. This subarea also has the highest average elevation of all subareas. Geology of the Hubbard and Terror Creeks subarea consists of landslide deposits, Wasatch Formation, and the upper part of the Mesaverde Formation, including the Ohio Creek member at the top, and Mancos The Hubbard and Terror Creeks subarea also has a lower average elevation than the Muddy Creek and Anthracite Creek subareas. The North Fork Gunnison River subarea primarily represents the mainstem of that river and was delineated as a subarea to compare the water quality of the North Fork Gunnison River to the other subareas. The geology of this subarea consists mostly of alluvial deposits.

To test the validity of subdividing the area, a statistical analysis of the average concentrations for each major dissolved and trace element constituent from the 19 stations was completed. Using the Duncan's multiple-range test, results showed that, generally, the stations were grouped into the four major subareas shown in figure 3, and the four subareas had statistically different water-quality characteristics, significant at the 95-percent level.

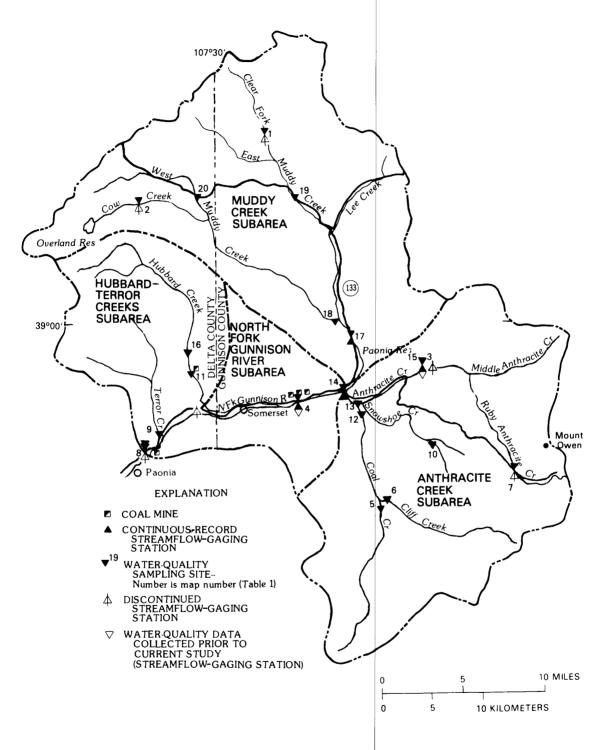


Figure 3.--Location of subareas.

# Dissolved Solids and Specific Conductance

Dissolved solids is the total concentration of dissolved material in water (Hem, 1970). Dissolved-solids concentrations often are used to describe general water quality and compare the water quality of different areas. At many sites, relations between streamflow and dissolved-solids concentrations can be developed for estimating dissolved-solids concentration from streamflow. Relations of this type for the four subareas and the entire study area are presented in table 3. As demonstrated by the correlation coefficients (r-values) in table 3, only the Hubbard and Terror Creeks subarea, and possibly the North Fork Gunnison River subarea, have a relation that adequately provides a dissolved-solids concentration estimate from streamflow.

Table 3.--Regression equations to predict dissolved-solids concentration from streamflow

Subarea name	Ex- ponent <sup>1</sup> B	Coeffi- cient <sup>1</sup> A	Corre- lation coeffi- cient (r)	Root mean square error (percent)	Number of data pairs (n)	
Muddy Creek						
subarea	-0.058	147	-0.27	40.0	77	0.0093
Anthracite Creek						•
subarea	057	71.1	37	21.8	54	.0031
Hubbard and Terror						
Creeks subarea	162	196	85	25.2	30	.0001
North Fork Gunnison						
River subarea	117	172	64	21.6	70	.0001
Total Area	109	151	49	41.5	234	.0001

<sup>&</sup>lt;sup>1</sup>The equation is:  $S = A(Q)^B$ , where S = dissolved-solids concentration, in milligrams per liter; and Q = streamflow, in cubic feet per second.

Plots of the streamflow dissolved-solids concentration relations in table 3 are shown in figure 4. Two important aspects of the area are demonstrated in figure 4: (1) Water in the Anthracite Creek subarea has a smaller dissolved-solids concentration than water in the other subareas; and (2) Muddy Creek and Anthracite Creek have a direct effect on the water quality of the North Fork Gunnison River. Approximately 91 percent of the mean-annual streamflow measured for the North Fork Gunnison River at the most downstream site (fig. 2, site 8) is contributed by the Muddy Creek and Anthracite Creek subareas. As the mean-annual streamflow from the Anthracite Creek subarea is approximately 50 percent larger than the mean-annual streamflow from the Muddy Creek subarea, the water quality in the North Fork Gunnison River would be

expected to resemble more closely the water quality from the Anthracite Creek subarea. However, this is not the case as shown in figure 4. At smaller streamflows (larger dissolved-solids concentrations), the value of the dissolved-solids concentration in the North Fork Gunnison River is nearer to the dissolved-solids concentration in the Muddy Creek subarea. In addition, the mean value of the dissolved-solids concentration (table 4) in the North Fork Gunnison River subarea is between the mean value of the dissolved-solids concentration of the Muddy Creek subarea and the mean value of the dissolved-solids concentration of the Anthracite Creek subarea.

Paonia Reservoir is a probable factor causing the similar dissolved-solids concentration in the North Fork Gunnison River subarea to the dissolved-solids concentration of the Muddy Creek subarea at lesser streamflows. As with most reservoirs, the effects of Paonia Reservoir on streamflow are to decrease peak flows and increase low flows by storing flood flows, and slowly releasing the stored water. Increasing the percentage of streamflow in the North Fork Gunnison River during smaller streamflows from Muddy Creek probably causes the dissolved-solids concentration in the North Fork Gunnison River to be nearer the concentration in Muddy Creek than the concentration in Anthracite Creek during those smaller streamflows. Other effects of Paonia Reservoir are discussed in a later section.

The largest dissolved-solids concentrations in the study area are in the Hubbard and Terror Creeks subarea (table 4). These relatively large concentrations in the Hubbard and Terror Creek subarea are mainly caused by the saline sedimentary formations in the basins. The mean dissolved-solids concentration for the total area is 97 mg/L.

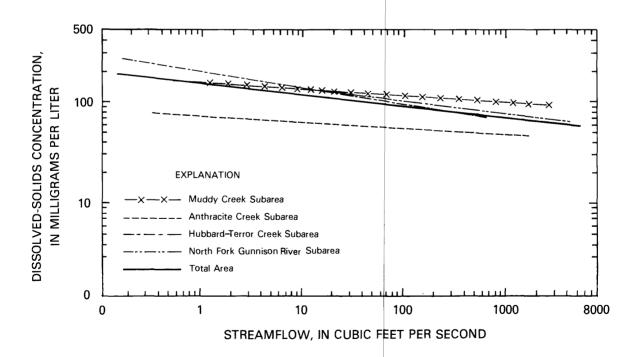


Figure 4.--Relation between streamflow and dissolved-solids concentration by subarea.

[mg/L, milligrams per liter; p, probability of obtaining a larger value of r with n pairs of randomly related data]Table 4.--Dissolved-solids concentration statistics by subarea

					Regress	Regression statistics <sup>1</sup>	stics1		
				Stan-		Corre- lation	Root	Number	
		Mini-	Maxi-	dard	Coeffi-	coeffi-	square	of data	Proba-
Subarea	Mean	<b>wnw</b>	mnm	devia-	cient	cient	error	pairs	bility
name	(mg/L)	(mg/L)	(mg/L)	tion	( <b>A</b> )	(r)	(percent) (n)	(u)	(b)
Muddy Creek									
subarea	122	37	233	46.23	0.615	0.92	18.22	78	0.0001
Anthracite Creek									
subarea	55	34	87	13.55	.717	.62	10.67	63	.0001
Hubbard and Terror									
Creeks subarea	133	29	337	67.48	.645	.92	26.13	33	.0001
North Fork Gunnison									
River subarea	91	51	191	26.74	.589	. 82	15.25	73	.0001
Total area	26	34	337	48.51	.621	.93	17.80	250	.0001

<sup>1</sup>Regression statistics represent dissolved solids as the dependent variable and specific conductance as the independent variable. The equation is: S = KA, where S = dissolved-solids concentration, in milligrams per liter; K = specific conductance, in microsiemens per centimeter at 25 degrees Celsius; and A = a coefficient. Another method for estimating dissolved-solids concentration is to develop relations between dissolved solids and the specific conductance of the water. Relations of this type for this area are presented in table 4. Hem (1970) states that the equation for this relation is:

$$S = KA; (1)$$

where

S = dissolved-solids concentration, in milligrams per liter;

K = specific conductance, in microsiemens per centimeter at 25 degrees Celsius; and

A = a coefficient.

Based on the generally good correlation shown in table 4, the equations appear adequate for estimating dissolved-solids concentration from specific conductance in the study area. These relations are plotted in figure 5 and show similar relations for all subareas. The relatively small values of specific conductance for the Anthracite Creek subarea also are shown in figure 5.

Specific conductance values for the subareas and for the total area are summarized in table 5. Values for the statistical relation to estimate specific conductance from streamflow also are presented in table 5. As with the relations for estimating dissolved solids from streamflow, only the Hubbard and Terror Creeks subarea relation has good correlation (r = -0.89) to indicate an adequate relation for predicting specific conductance from streamflow. The correlation for the North Fork Gunnison River (r = -0.66) suggests that this equation also may be useful for estimating specific conductance from streamflow.

## Major Dissolved Constituents

For each subarea, the mean, minimum, maximum, and standard deviation for dissolved-calcium concentration are shown in table 6. The mean concentration of dissolved calcium in the North Fork Gunnison River subarea is quite similar to the average of the mean concentrations of the Muddy Creek and Anthracite Creek subareas. Values for the regression equations to predict dissolved calcium concentration from specific conductance also are presented in table 6 for each subarea and for the total area. These equations are plotted in figure 6. The equation for the total area, with an r-value of 0.94, appears to adequately predict dissolved-calcium concentration for any site within the study area, especially at smaller conductance values. However, at larger conductance values, the total area equation would slightly overpredict dissolved-calcium concentration for sites in all subareas except the Muddy Creek subarea, where the equation would underpredict (figure 6).

The mean, minimum, maximum, and standard deviation for dissolved-magnesium concentrations are presented in table 7 and the same values for sodium are in table 8. For both constituents, are in the Hubbard and Terror Creeks subarea, and the smallest mean concentrations are in the Anthracite Creek subarea. Regression-equation coefficients to estimate the constituent from specific-conductance values also are in tables 7 and 8. These equations are plotted in figure 7 for magnesium and in figure 8 for sodium. From table 7 and figure 7, all subareas have similar prediction equations for magnesium concentration, except for the Hubbard and Terror Creeks subarea, which has larger values and a steeper slope.

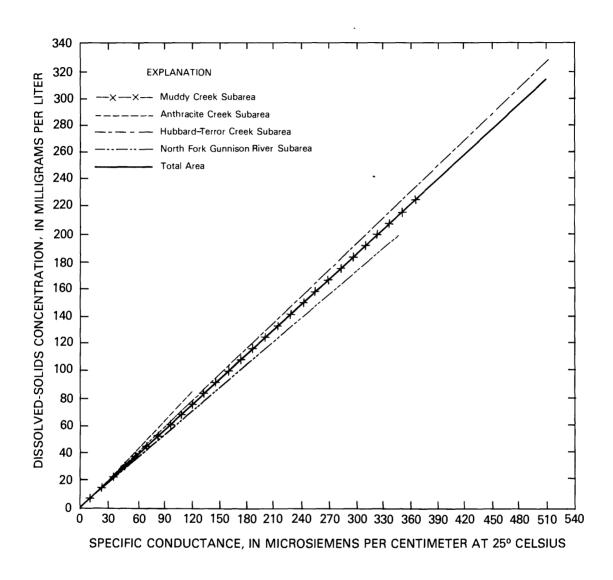


Figure 5.--Relation between specific conductance and dissolved-solids concentration by subarea.

[µS/cm, microsiemens per centimeter at 25 degrees Celsius; p, probability of obtaining a larger value Table 5.--Specific-conductance statistics by subarea of r with n pairs of randomly related data]

					Ř	Regression statistics <sup>1</sup>	statisti	.cs1		
							Corre-	Root		
				Stan-			lation	mean	Number	
		Mini-	Maxi-	dard	Expo-	Coeffi-	coeffi-	square	of data	Proba-
Subarea	Mean	mnm	mnm	devia-	nent	cient	cient	error	pairs	bility
name	(µS/cm)	(µS/cm) (µS/cm)	(mS/cm)	tion	þ	(A)	(r)	(percent)	(u)	(b)
Muddy Creek										
subarea	171	38	905	81.86	-0.052	203.56	-0.15	58.20	84	0.0845
Anthracite Creek										
subarea	7.1	29	155	27.55	051	84.61	22	39.16	<del>7</del> 9	.0483
Hubbard and Terror										
Creeks subarea	190	73	510	120.00	224	318.30	89	28.65	30	.0001
North Fork Gunnison										
River subarea	152	53	369	55.10	157	348.63	99	27.89	71	.0001
I										
Total area	144	29	510	83.00	099	197.35	34	59.82	255	.0001

<sup>1</sup>Regression statistics represent specifc conductance as the dependent variable, and streamflow as the independent variable. The equation is: Specific conductance, in microsiemens per centimeter at 25 degrees Celsius =  $A(Q)^b$ , where Q = streamflow, in cubic feet per second.

[mg/L, milligrams per liter; p, probability of obtaining a larger value of r with n pairs of randomly related data] Table 6. -- Calcium concentration statistics by subarea

		Mini-	Maxi-	Standard			Correla- tion coeffi-	Root mean square	Number of data Prob-	Prob-
Subarea name	Mean (mg/L)	mum (mg/L)	mum (mg/L)	devia- tion	Slope 1	Inter- cept <sup>1</sup>	cient (r)	error (mg/L)	pairs ability (n) (p)	ability (p)
Muddy Creek										
subarea	29	6.7	57	12.87	0.141	2.72	0.94	4.54	82	0.0001
Anthracite Creek										
subarea	11	5.0	23	3.32	.106	3.11	06.	1.41	73	.0001
Hubbard and Terror										
Creeks subarea	22	9.8	62	12.46	.100	3.23	76.	3.14	33	.0001
North Fork Gunnison	r									
River subarea	19	10.0	37	5.48	.088	5.86	.87	2.70	4/	.0001
Total area	20	5.0	62	11.59	.123	2.40	.94	4.43	271	.0001

equation is: Calcium concentration, in milligrams per liter = slope × specific conductance + intercept.  $^1$ Calcium is the dependent variable, and specific conductance is the independent variable. The

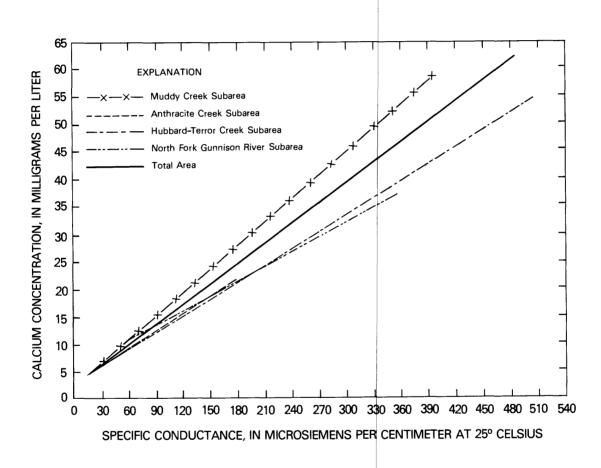


Figure 6.--Relation between specific conductance and calcium concentration by subarea.

 $[mg/L,\ milligrams\ per\ liter;\ p,\ probability\ of\ obtaining\ a\ larger\ value\ of\ r\ with\ n\ pairs\ of\ randomly\ related\ data]$ Table 7.--Magnesium concentration statistics by subarea

ខ	Mean	Mini- mum	Maxi- mum	Standard devia-	-	Inter-	Correla- tion coeffi- cient	Root mean square error	Number of data pairs	Prob- ability
name Muddv Creek	(mg/L)	(mg/L)	(mg/r)	tion	Slope	cept	(r)	(mg/r)	1	(d)
subarea	4.8	1.6	6.6	2.18	0.023	0.45	0.92	0.87	85	0.0001
Subarea Subard and Terror	1.8	∞.	4.1	.83	.025	04	.85	44.	73	.0001
nubbalu and itiloi Creeks subarea North Fork Gunnison	6.1	2.8	23	4.28	.032	03	.91	1.80	33	.0001
River subarea	3.5	1.8	11	1.53	.024	90	.85	.82	74	.0001
Total area	3.8	∞.	23	2.61	.027	20	.91	1.07	271	.0001

 $^1$ Magnesium is the dependent variable, and specific conductance is the independent variable. The equation is: Magnesium concentration, in milligrams per liter = slope  $\times$  specific conductance + intercept.

[mg/L, milligrams per liter; p, probability of obtaining a larger value of r with n pairs of randomly related data]Table 8.--Sodium concentration statistics by subarea

							Correla-	Root		
							tion	mean	Number	
		Mini-	Maxi-	Standard			coeffi-	square	of data	Prob-
Subarea	Mean	mnm	mnm	devia-		Inter-	cient	error	pairs	ability
name	(mg/L)	(mg/L)	(mg/L)	tion	Slope1	cept1	(r)	$(mg/\Gamma)$	(u)	(b)
Muddy Creek									i	
subarea	6.5	1.1	18	3.43	0.029	1.03	0.73	2.29	82	0.0001
Anthracite Creek										
subarea	3.7	1.5	8.9	1.55	.020	2.33	.33	1.47	73	.0021
Hubbard and Terror										
Creeks subarea	13	3.4	70	8.94	.062	7.4	.84	4.89	33	.0001
North Fork Gunnison										
River subarea	7.0	2.9	<u> </u>	3.24	970.	.210	77.	2.07	74	.0001
Total area	9.9	1.1	40	4.91	.042	09.	74.	3.31	271	.0001

Sodium concentration, in milligrams per liter = slope × specific conductance + intercept. 1Sodium is the dependent variable, and specific conductance is the independent variable. The equation is:

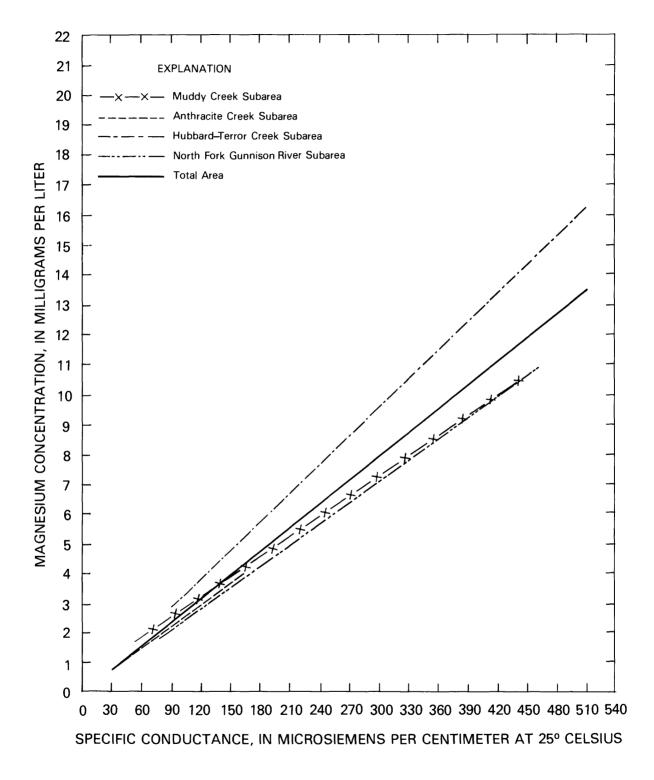


Figure 7.--Relation between specific conductance and magnesium concentration by subarea.

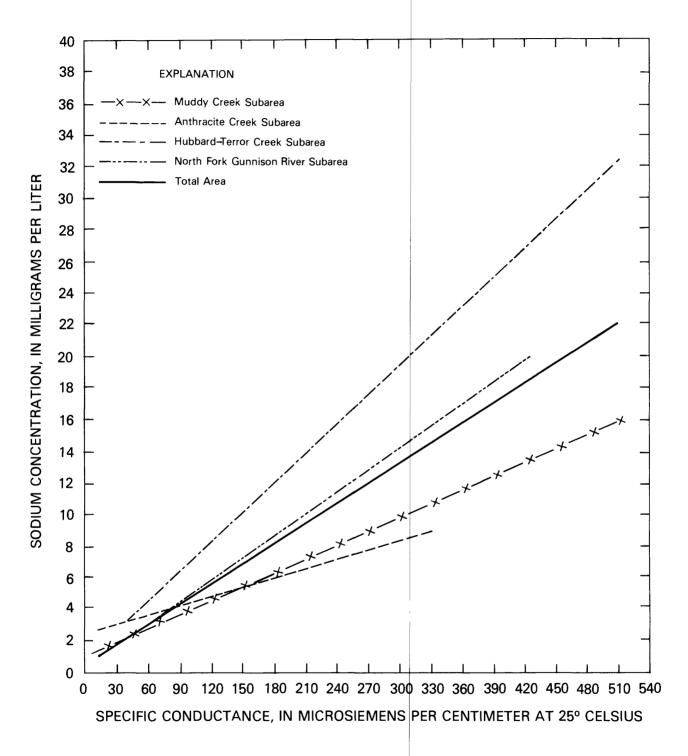


Figure 8.--Relation between specific conductance and sodium concentration by subarea.

Compared to figure 7, equations to predict sodium concentration from specific conductance plotted in figure 8 are more varied. However, for sodium concentration, the Hubbard and Terror Creeks subarea again has larger values and a steeper slope.

The mean, minimum, maximum, and standard deviation for alkalinity are presented in table 9. Hem (1970) defines alkalinity "as the capacity of the solution to neutralize acid." In most natural water, alkalinity is produced mostly by dissolved carbonate and bicarbonate ions. Measured alkalinity concentrations in the study area ranged from 7 mg/L to 220 mg/L. Both maximum and minimum alkalinity concentrations for the study area were from the Hubbard and Terror Creeks subarea.

Regression-equation values to predict alkalinity from specific conductance for each subarea also are in table 9. These equations are plotted in figure 9, which indicates that, generally, alkalinity concentrations were larger in the Muddy Creek and Hubbard and Terror Creeks subareas, and smaller in the Anthracite Creek and North Fork Gunnison River subareas.

Mean minimum, maximum, and standard-deviation values are presented in table 10 for potassium, table 11 for chloride, table 12 for silica, and table 13 for sulfate. These constituents have small mean values, with the possible exception of sulfate values (table 13) in the Hubbard and Terror Creeks and North Fork Gunnison River subareas; these are discussed in later sections. Regression-equation values to estimate each constituent from specific conductance also are presented in tables 10 through 13. However, as shown in these tables, generally the corelation coefficient is small, and use of many of the equations would provide questionable results. For equations represented in tables 10 through 13, equations with a small slope value indicate that the constituent has little relation with specific conductance. This especially is true for potassium (table 10) and chloride (table 11). For these constituents, the best estimate would be the mean concentration, with a range of plus and minus the standard deviation.

# Suspended Sediment and Trace Elements

Within normal pH ranges, trace elements often are attached to sediment particles instead of dissolved, and a relation often can be developed between suspended-sediment and trace-element concentrations. Suspended-sediment concentration data are presented in table 14 for the subareas in the study area. The average suspended-sediment concentration for the entire area is approximately 214 mg/L, with the Anthracite Creek and North Fork Gunnison River subareas having nearly equal concentrations, with an average of 126 mg/L. However, the Muddy Creek subarea has an average suspended-sediment concentration nearly 3 times larger (361 mg/L). The source of the larger suspended-sediment concentration is unknown, but probably is related to landslides and easily erodible soils.

Mean daily suspended-sediment discharges for the subareas also are included in table 14. The largest mean daily suspended-sediment discharge occurred at the station with the largest drainage area, the North Fork Gunnison River above Paonia (fig. 2, site 8) site (North Fork Gunnison River

 $[mg/L,\ milligrams\ per\ liter;\ p,\ probability\ of\ obtaining\ a\ larger\ value\ of\ r\ with\ n\ pairs\ of\ randomly\ related\ data]$ Table 9.--Alkalinity concentration statistics by subarea

Subarea name	Mean (mg/L)	Mini- mum (mg/L)	Maxi- mum (mg/L)	Standard devia- tion	Slope1	Inter- cept <sup>1</sup>	Correla- tion coeffi- cient (r)	Root mean square error (mg/L)	Number of data pairs (n)	Prob- ability (p)
Muddy Creek	,		,		,				,	
subarea	100	23	200	44.33	0.488	8.51	0.62	14.54	85	0.0001
Anthracite Creek										
subarea	36	18	69	10.59	.322	12.90	.84	5.68	73	.0001
Hubbard and Terror										
Creeks subarea	88	7.0	220	52.11	.412	9.83	.95	15.55	33	.0001
North Fork Gunnison	,									
River subarea	65	30	130	20.56	.322	17.05	.85	10.76	73	.0001
Total area	71	7.0	220	42.08	.455	5.17	76.	14.46	270	.0001

equation is: Alkalinity concentration, in milligrams per liter = slope × specific conductance + intercept.  $^1$ Alkalinity is the dependent variable, and specific conductance is the independent variable. The

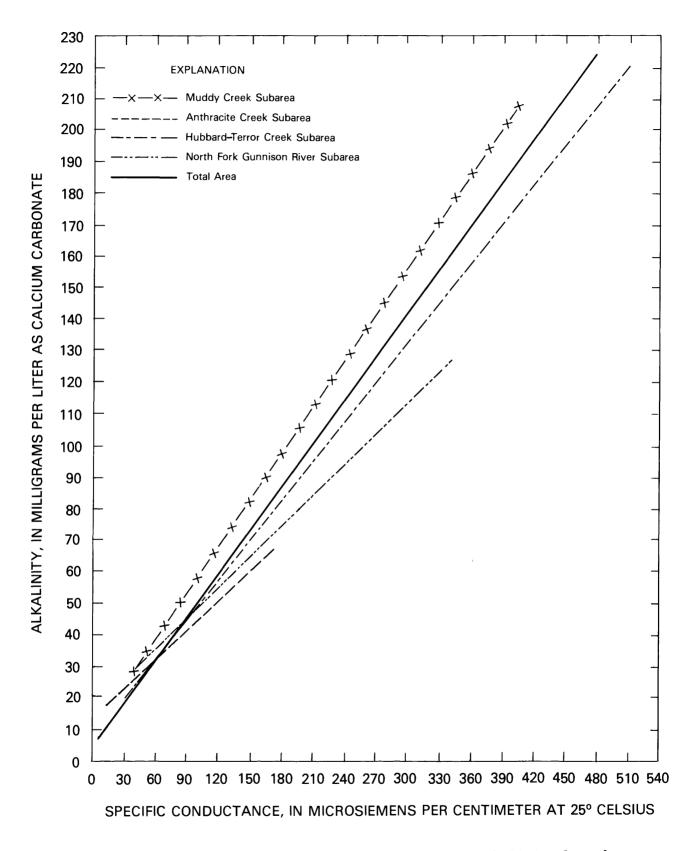


Figure 9.--Relation between specific conductance and alkalinity by subarea.

[mg/L, milligrams per liter; p, probability of obtaining a larger value of r with n pairs of randomly related data]Table 10.--Potassium concentration statistics by subarea

Subarea name	Mean (mg/L)	Mini- mum (mg/L)	Maxi- mum (mg/L)	Standard devia- tion	Slope 1	Inter- cept <sup>1</sup>	Correlation coefficient (r)	Root mean square error (mg/L)	Number of data pairs (n)	Prob- ability (p)
Muddy Creek subarea	1.1	0.5	2.4	0.421	0.003	0.58	0.62	0.33	85	0.0001
Anthracite Creek subarea	7.	·.3	3.1	. 487	900.	.29	.30	97.	73	.0050
Hubbard and Terror Creeks subarea	1.7	φ.	5.3	.885	.005	.78	99.	99.	33	.0001
Norch Fork Gumison River subarea	8.	4.	1.7	. 288	.002	67.	77.	.26	7,4	.0001
Total area	1.0	÷.	5.3	.590	700.	07.	.62	97.	271	.0001

equation is: Potassium concentration, in milligrams per liter = slope × specific conductance + intercept. <sup>1</sup>Potassium is the dependent variable, and specific conductance is the independent variable. The

[mg/L, milligrams per liter; p, probability of obtaining a larger value of r with n pairs of randomly related data] Table 11.--Chloride concentration statistics by subarea

Muddy Creek subarea 1.3 0.5 Anthracite Creek	fini- Maxi- num mum ng/L) (mg/L)	Standard devia- tion	Slope1	Inter- cept <sup>1</sup>	tion coeffi- cient (r)	mean square error (mg/L)	Number of data pairs (n)	Prob- ability (p)
Anthracite treek	8.6	0.925	0.003	0.79	0.22	0.89	85	0.0258
subarea .8 .2	3.4	.661	700.	.53	. 14	.65	73	.1181
Creeks subarea 1.9 .1	5.0	1.300	.007	64.	.68	96.	33	.0001
River subarea 1.6 .5	3.7	606.	.010	.15	.58	74.	74	.0001
Total area 1.3 .1	8.6	.980	900.	.54	. 48	98.	271	.0001

equation is: Chloride concentration, in milligrams per liter = slope × specific conductance + intercept. <sup>1</sup>Chloride is the dependent variable, and specific conductance is the independent variable. The

[mg/L, milligrams per liter; p, probability of obtaining a larger value of r with n pairs of randomly related data]Table 12.--Silica concentration statistics by subarea

Subarea name	Mean (mg/L)	Mini- mum (mg/L)	Maxi- mum (mg/L)	Standard devia- tion	Slope1	Inter- cept 1	Correla- tion coeffi- cient (r)	Root mean N square o error (mg/L)	umber f data pairs (n)	Prob- ability (p)
Muddy Creek	0	6 4	7,	1 60	0.00	7 11	7, 0	71 1	0	1000
subarea Anthracite Creek	0.0	7.0	<b>+</b> 1	1.00	0.013	77./	, . ,	1.14		0.0001
subarea	0.6	2.7	15	2.97	.024	7.26	.20	2.92	73	.0529
Hubbard and Terror Creeks subarea	16	12	93	13.50	016	19.51	09	13.56	33	.4104
North Fork Gunnison River subarea	8.8	3.9	11	.95	.007	7.80	.36	88.	74	8000.
Total area	10	2.7	93	5.68	600.	8.79	.14	5.64	271	.0179

Silica concentration, in milligrams per liter = slope × specific conductance + intercept. The 1Silica is the dependent variable, and specific conductance is the independent variable. equation is:

[mg/L, milligrams per liter; p, probability of obtaining a larger value of <math>r with n pairs of randomly related data] Table 13. -- Sulfate concentration statistics by subarea

Subarea name	Mean (mg/L)	Mini- mum (mg/L)	Maxi- mum (mg/L)	Standard devia- tion	Slope1	Inter- cept <sup>1</sup>	Correla- tion coeffi- cient (r)	Root mean square error (mg/L)	Number of data pairs (n)	Prob- ability (p)
Muddy Creek subarea	6.7	3.6	13	1.81	0.005	5.79	0.20	1.78	85	0.0397
Anthracite Creek subarea	6.8	2.5	13	2.39	.047	3.45	.54	2.01	73	.0001
Hubbard and Terror Creeks subarea	17	5.0	74	14.13	660.	-2.17	78.	7.02	33	.0001
North Fork Gunnison River subarea	10	9.	04	6.13	890.	. 23	09.	4.92	73	.0001
Total area	9.0	9.	74	6.99	.042	2.94	.52	5.99	270	.0001

 $^1$ Sulfate is the dependent variable, and specific conductance is the independent variable. The tion is: Sulfate concentration, in milligrams per liter = slope  $\times$  specific conductance + intercept. equation is:

Table 14.--Suspended-sediment concentration statistics by subarea [mg/L, milligrams per liter]

Subarea name	Mean (mg/L)	Minimum (mg/L)	Maximum (mg/L)	Standard deviation	Mean daily suspended- sediment discharge (tons per day)
Muddy Creek subarea Anthracite Creek	361	4	5,790	865.8	362
subarea	122	3	1,660	268.8	174
Hubbard and Terror Creeks subarea North Fork Gunnison	201	3	2,150	394.3	78
River subarea	130	1	1,590	274.7	693
Total area	214	1	5,790	561.4	355

subarea), with a mean daily suspended-sediment discharge of 1,690 ton/d. The smallest mean suspended-sediment discharge occurred at the Grouse Spring Creek near Marcellina Mountain (fig. 2, site 10) site (Anthracite Creek subarea), with a mean daily suspended-sediment discharge of 3.9 ton/d. This station has the smallest drainage area of all the sites where analyses were made.

The mean suspended-sediment concentrations and discharges were computed only from the 2 years of synoptically collected data. Subarea and total area mean values are the mean value of all data within that area.

Concentration statistics for total iron, manganese, and zinc for the subareas are presented in tables 15, 16, and 17. For these three trace elements, the Muddy Creek subarea had the largest mean concentrations, and the Hubbard and Terror Creeks subareas had the smallest mean concentrations. The most probable explanation for the Muddy Creek subarea having the largest trace-element concentrations is that it had the largest mean suspended-sediment concentration (table 14). Total-iron concentrations (table 15) appear large compared to the other trace elements analyzed; the cause for the large concentration is unknown, but it is most likely related to the natural large iron content in the area.

Regression-equation values to predict trace-element concentrations from suspended-sediment concentrations also are in table 15 for iron, table 16 for manganese, and table 17 for zinc. Prediction equation values for total-iron concentrations for the Muddy Creek and Anthracite Creek subareas are similar in slope and intercept (table 15); however, the Muddy Creek subarea values had maximum concentrations over 3 times larger than the Anthracite Creek subarea concentrations.

[µg/L, micrograms per liter; p, probability of obtaining a larger value of r with n pairs of randomly related data] Table 15.--Total-iron concentration statistics by subarea

	:	Mini-	Maxi-	Standard			Correla- tion coeffi-	Root mean square		Prob-
Subarea name	Mean (µg/L)	$\mu$ mum $(\mu g/\Gamma)$	mum (µg/L)	devia- tion	Slope1	Inter- cept <sup>1</sup>	cient (r)	error (µg/L)	pairs (n)	ability (p)
Muddy Creek										
subarea	10,700	077	71,000	14,260	9.686	2,490	0.89	6,404	24	0.0001
Anthracite Creek										
subarea	6,500 1,600	1,600	18,000	5,510	10.490	1,310	98.	2,827	11	. 0002
Hubbard and Terror										
Creeks subarea	5,700	1,300	11,000	3,450	4.155	3,810	.59	2,781	13	.0151
North Fork Gunnison	•									
River subarea	7,700 1,700	1,700	15,000	3,890	5.522	4,340	.63	3,013	7	.0543
									,	
Total area	8,200	077	71,000	9,960	9.279	2,210	. 88	7,806	26	.0001

 $^1\mathrm{Total}$  iron is the dependent variable, and suspended-sediment concentration is the independent variable. The equation is: Total-iron concentration, in micrograms per liter = slope  $\times$  suspendedsediment concentration + intercept. Statistics for total-lead and total-cadmium concentrations for the subareas are presented in table 18 and in table 19. Because correlation coefficients for the equations to estimate these trace-elements concentrations from suspended sediment generally are small, plots of these equations are not given. The largest mean total-lead concentration was found in the North Fork Gunnison River subarea; the largest maximum total-lead concentration was found in the Muddy Creek subarea (table 18).

As with total lead, the largest total-cadmium concentration analyzed was in the Muddy Creek subarea (table 19). In both tables 18 and 19, a value of 1  $\mu$ g/L indicates the sample had trace-element concentration at or below detection limits.

# Effects of Paonia Reservoir

Paonia Reservoir is located on Muddy Creek just upstream from the confluence of Muddy Creek and Anthracite Creek, where the combined flows become the North Fork Gunnison River. The reservoir had some effect on each water-quality constituent measured in the study from sites upstream (site 17) and downstream (site 14). These effects are summarized in table 20. Mean suspended-sediment discharge leaving the reservoir decreased by nearly 63 percent, with a mean suspended-sediment discharge of 1,395 ton/d entering the reservoir, but only 518 ton/d leaving the reservoir. These mean values only are from the two years of synoptically collected data.

This decreased suspended-sediment concentration leaving the reservoir had a major effect on the trace elements that attach to sediments. The mean trace-element concentration decreased by an average of nearly 67 percent downstream from the reservoir as compared to upstream (table 20).

Concentrations of major ions in the dissolved phase also decreased as the water flowed through the reservoir (table 20). Sulfate decreased the least (1.6 percent) and chloride decreased the most (approximately 22 percent) downstream from the reservoir as compared to upstream from the reservoir. Although mean chloride concentrations decreased by nearly 22 percent as water moved through the reservoir, the actual decrease was only 0.4 mg/L. This decrease, as for many of the major-dissolved constituents, probably is related to the reservoir storage of low-concentration runoff. No samples were collected in the reservoir.

## Effects of Coal Mining

General coal-mining effects on water quality have been known for some time. Pyritic materials, generally iron pyrite  $(FeS_2)$  are exposed to the atmosphere and water, and react to form ferrous sulfate  $(FeSO_4)$  and sulfuric acid  $(H_2SO_4)$ . This breakdown of pyrite usually increases the concentration of sulfate, hydrogen ions, and iron in the water. Increased concentration of hydrogen ions results in low pH, which is a common characteristic of many coal-mine drainage waters (Biesecker and George, 1966). Reaction of this low-pH water with carbonate materials decreases acidity (increases pH) and increases dissolved-solids concentration. Thus, a measure of coal-mining effects on water quality is provided by examining dissolved-solids concentration, sulfate concentration, and pH or alkalinity.

 $[\mu g/L,$  micrograms per liter; p, probability of obtaining a larger value of r with n pairs of randomly related data] Table 16. -- Total-manganese concentration statistics by subarea

Subarea name	Mean (µg/L)	Mini- mum (μg/L)	Maxi- mum (μg/L)	Standard devia- tion	Slope 1	Inter- cept <sup>1</sup>	Correla- tion coeffi- cient (r)	Root mean square error (µg/L)	Number of data pairs (n)	Prob- ability (p)
Muddy Creek subarea	380	10	1,600	416	0.260	165	0.82	238	24	0.0001
Anthracite Creek subarea	200	07	510	155	.321	42.2	.95	7.67	11	.0001
Hubbard and Terror Creeks subarea	140	07	280	76.4	.0840	103	.52	6.49	13	.0302
Norch Fork Gumison River subarea	190	10	200	141	.185	88.7	.56	117	8	.0688
Total area	260	10	1,600	300	.260	95.3	.81	117	56	.0001

 $^1\mathrm{Equation}$  is: Total-manganese concentration, in micrograms per liter = slope imes suspended-sediment concentration + intercept.

 $[\mu g/L,$  micrograms per liter; p, probability of obtaining a larger value of r with n pairs of randomly related data] Table 17. -- Total-zinc concentration statistics by subarea

Subarea name	Mean (µg/L)	Mini- mum (µg/L)	Maxi- mum (μg/L)	Standard devia- tion	Slope 1	Inter- cept <sup>1</sup>	Correlation coefficient (r)	Root mean square error (µg/L)	Number of data pairs (n)	Prob- ability (p)
Muddy Creek subarea	06	0	580	112	0.077	21.3	0.91	48.2	24	0.0001
Anthracite Creek subarea	50	20	96	23.9	.038	34.4	.71	16.9	11	.0063
Hubbard and Terror Creeks subarea	50	20	130	27.8	.048	25.9	.91	11.7	13	.0001
North Fork Gunnison River subarea	09	0	100	31.3	.047	28.7	89.	23.1	∞	.0276
Total area	70	0	580	76.5	.073	19.3	. 89	34.7	26	.0001

The equation is: Total-zinc concentration, in micrograms per liter = slope × suspended-sediment concentration + intercept.

[µg/L, micrograms per liter; p, probability of obtaining a larger value of r with n Table 18. -- Total-lead concentration statistics by subarea pairs of randomly related data]

Subarea name	Mean (µg/L)	Mini- mum (µg/L)	Maxi- mum (μg/L)	Standard devia- tion	Inter- Slope <sup>1</sup> cept <sup>1</sup>	Inter- cept 1	Correlation coefficient (r)	Root mean square error (µg/L)	Number of data pairs a	Prob- ability (p)
Muddy Creek subarea	16	4	74	16.2	0.008	10.093	0.59	13.0	23	0.0011
Subarea  Hubbard and Terror	12	7	23	5.95	600.	7.473	99.	4.49	11	.0123
Creeks subarea	∞ .	21	17	4.78	.005	5.704	.52	4.08	13	.0320
River subarea	18	4	41	12.3	.013	11.243	.36	11.4	8	.1777
Total area	14	21	74	12.4	800.	8.816	.58	10.0	55	.0001

 $^1\mathrm{The}$  equation is: Total-lead concentration, in micrograms per liter = slope imes suspended-sediment concentration + intercept.

<sup>2</sup>A concentration value of 1 microgram per liter represents a value at or below laboratorydetection level.

 $[\mu g/L$ , micrograms per liter; p, probability of obtaining a larger value of r with n pairs of randomly related data] Table 19. -- Total-cadmium concentration statistics by subarea

Subarea name	Mean (µg/L)	Mini- mum (µg/L)	Maxi- mum (µg/L)	Standard devia- tion	Slope1	Inter- cept <sup>1</sup>	Correla- tion coeffi- cient (r)	Root mean square error (µg/L)	Number of data pairs a	Prob- ability (p)
Muddy Creek subarea	21	21	7	92.0	-0.0001	1.46	-0.10	0.78	24	0.5693
Anthracite Greek subarea	21	21	21	0.	00.	1.0	1.0	00.	11	.0001
hubbard and lerror Creeks subarea	21	21	7	.43	.0003	1.08	.26	.41	13	.1827
North Fork Gummison subarea	21	21	6	.73	.0007	1.02	.36	89.	<b>∞</b>	.1822
Total area	21	21	4	.61	000.	1.27	.10	.62	26	.8680

= slope × suspended-1The equation is: Total-cadmium concentration, in micrograms per liter sediment concentration + intercept.

<sup>2</sup>A concentration value of 1 microgram per liter represents a value at or below laboratory detection level.

[ $\mu g/L$ , micrograms per liter;  $\mu S/cm$ , microsiemens per centimeter at 25 degrees Celsius; ft $^3/s$ , cubic feet per second; mg/L, milligrams per liter; CaCO $_3$ , calcium carbonate] Table 20. -- Effects of Paonia Reservoir on water quality of Muddy Creek

		Upstream from	from reservoir	ir		Downstream	Downstream from reservoir	oir	
Constituent or property	Mean	Minimum	Maximum	Standard deviation	Mean	Minimum	Maximum	Standard deviation	Percent decrease in mean
Streamflow (ft <sup>3</sup> /s)	769	58	2,500	738.1	678	31	2,520	693.1	11.8
Dissolved solids (mg/L)	122	79.4	181	31.2	106	84.5	135	16.5	13.3
Specific conductance (µS/cm)	191	120	305	56.2	159	107	215	33.1	16.8
Suspended sediment (mg/L)	862	58	3,660	1,090	132	4	372	146	84.7
Calcium (mg/L)	28	19	77	7.82	24	19	31	4.00	14.3
Magnesium (mg/L)	4.7	3.0	7.6	1.42	3.9	3.0	5.3	.77	17.0
Sodium (mg/L)	7.5	3.9	11	2.20	6.9	4.5	13	2.48	8.0
Chloride $(mg/L)$	1.8	. 70	8.6	1.92	1.4	∞.	2.8	.52	22.2
Sulfate $(mg/L)$	6.2	5.0	0.6	1.00	6.1	5.0	10	1.27	1.6
Alkalinity (mg/L as CaCO <sub>3</sub> )	100	62	161	28.9	85	69	117	14.9	15.0
Total iron (µg/L)	10,900	3,300	24,000	8,750	3,540	620	7,200	2,450	67.7
Total manganese (µg/L)	067	150	1,100	410.0	150	20	480	172	7.69
Total zinc (µg/L)	80	30	140	43.8	30	20	20	13.8	62.5
Total lead (µg/L)	22	9	54	16.9	7	4	15	3.83	68.2

#### Hubbard Creek

Water-quality sampling sites at lower Hubbard Creek near Bowie (fig. 2, site 11) and upper Hubbard Creek near Bowie (fig. 2, site 16) were established in an attempt to detect any changes in water quality that might be attributed to a coal mine located between these two sites. The relation between dissolved solids and streamflow for the two sites on Hubbard Creek are plotted in figure 10. Lower Hubbard Creek (downstream from mining) had a slightly larger concentration of dissolved solids at smaller streamflows than the upper Hubbard Creek site. However, the increase was mining, but may have been caused by downstream effects such as evaporation and ground-water inflow.

For the lower Hubbard Creek site, the regression equation plotted in figure 10 is:

DS = 
$$244.91 \text{ Q}^{-0.195}$$
; and r =  $0.92$ . (2)

For the upper Hubbard Creek site, the regression equation is:

DS = 
$$190.11(Q)^{-0.139}$$
; and r =  $0.69$ ; (3)

where

DS = dissolved-solids concentration, in milligrams per liter; and Q = streamflow, in cubic feet per second.

Mean dissolved-solids concentration for the upper Hubbard Creek site was 128 mg/L and 137 mg/L for the lower Hubbard Creek site, a 7-percent increase.

Sulfate concentration histograms for the two sites on Hubbard Creek are shown in figure 11. Mean sulfate concentration at the lower Hubbard Creek site was 14.9 mg/L and, at the upper Hubbard Creek site was 11.1 mg/L, a 36.4 percent increase. Although the lower Hubbard Creek site (downstream from mining) had some sulfate concentrations larger than the upper Hubbard Creek site (figure 11), it is unknown if the differences are because of coal mining or because of downstream effects from evaporation and ground-water inflow or a combination of the two. To determine the cause of the increased sulfate and dissolved-solids concentrations at the lower Hubbard Creek site, more detailed sampling would be required.

Generally, acid coal-mine drainage into streams causes alkalinity to decrease. However, in the Hubbard Creek drainage, this was not observed. The mean alkalinity concentration, as  $CaCO_3$ , upstream from mining was 82 mg/L and downstream from mining was 95 mg/L, an increase of 15.9 percent. These data, and the discussion of dissolved-solids and sulfate concentration, suggest that, in Hubbard Creek, coal mining had no appreciable effects on water quality. To obtain more definitive results, however, more detailed sampling would be needed, as discussed earlier.

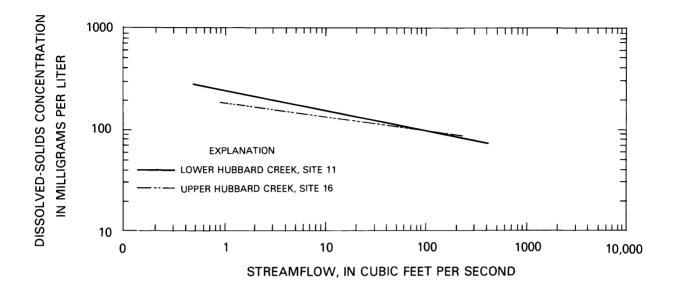


Figure 10.--Relation between streamflow and dissolved-solids concentration for upper and lower Hubbard Creek near Bowie.

## North Fork Gunnison River

Most coal mining in the study area occurs along the banks of the North Fork Gunnison River. Because of an absence of bridges over the river and generally high streamflow, no water-quality sampling site could be established on the North Fork Gunnison River upstream from coal mining. However, water-quality sampling sites are on Muddy Creek (fig. 2, site 14) and Anthracite Creek (fig. 2, site 13), upstream from the convergence of these two streams, where they form the North Fork Gunnison River. Using an equation that weights the concentration by flow, estimates for concentrations of dissolved solids, sulfate, and alkalinity were made for the point of convergence of the two streams. The equation used was:

$$c_3 = \frac{Q_1 C_1 + Q_2 C_2}{Q_1 + Q_2} \quad , \tag{4}$$

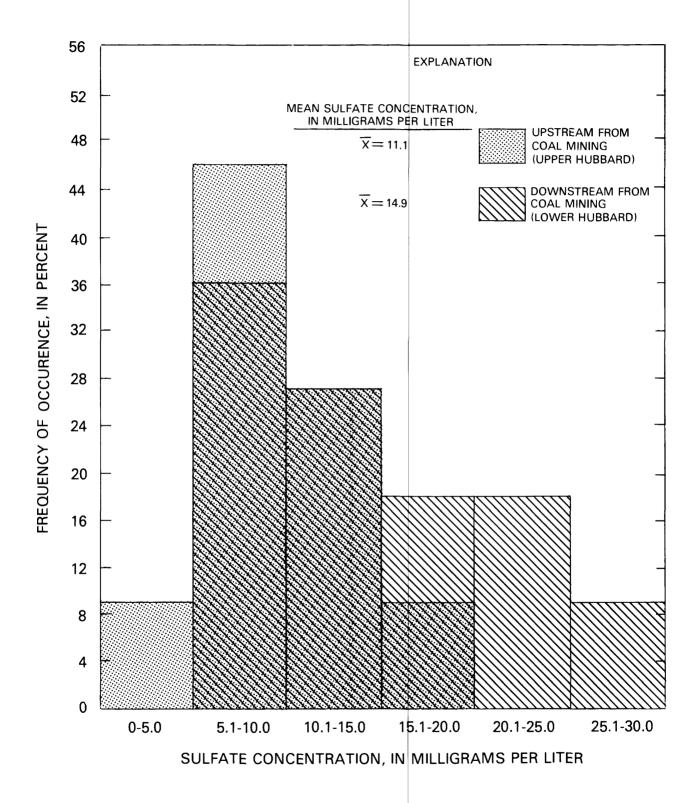


Figure 11.--Frequency of occurrence of sulfate concentrations for upper and lower Hubbard Creek, upstream and downstream from coal-mining activities.

#### where

- $\mathbf{Q}_1$  = streamflow for Muddy Creek below Paonia Reservoir (fig. 2, site 14), in cubic feet per second; and
- C<sub>1</sub> = concentration of the specific constituent for Muddy Creek
   below Paonia Reservoir (fig. 2, site 14); and
- Q<sub>2</sub> = streamflow for lower Anthracite Creek near Somerset (fig. 2, site 13), in cubic feet per second; and
- $C_3$  = the constituent concentration for the point of convergence of the two streams.

The equation was used to transfer constituent data to the convergence point, generally when samples were collected the same day at the two sites. The point of convergence of Muddy Creek and Anthracite Creek to form the North Fork Gunnison River is called the upper North Fork Gunnison River site in this report.

Mean dissolved-solids concentration for the North Fork Gunnison River above Paonia site was 102 mg/L and, for the upper North Fork Gunnison River site it was 75.6 mg/L. At smaller streamflows, dissolved-solids concentration downstream from the mining site (North Fork Gunnison River above Paonia) tend to be larger than for the estimated data at the upper North Fork Gunnison River site. Although these data could be an indication of coal-mining effects on water quality, the dissolved-solids concentration is small, and too few data are available to define the cause of this increase.

Frequency of occurrence of sulfate concentrations for the North Fork Gunnison River above Paonia and for the upper North Fork Gunnison River sites are in figure 12. As for dissolved-solids concentration, the larger sulfate concentrations were at the North Fork Gunnison River above Paonia site, downstream from mining. However, as for dissolved-solids concentration, mean sulfate concentrations were small, and too few data exist to determine the cause for the increase in sulfate concentrations at the site downstream from mining. As for Hubbard Creek, the increase in sulfate possibly is related to downstream effects discussed earlier. Although the mean sulfate concentration did increase more than 100 percent (from 7.0 to 14.2 mg/L) between the two sites, other dissolved-constituent concentrations also increased substantially. For example, chloride concentrations increased more than 70 percent through the same stream reach.

As for Hubbard Creek, there were no major effects on alkalinity concentrations in the North Fork Gunnison River resulting from coal mining. Mean alkalinity, as  $CaCO_3$ , was 57 mg/L for the upper North Fork Gunnison River site and was 70 mg/L for the North Fork Gunnison River above Paonia site, a downstream increase of 22.8 percent.

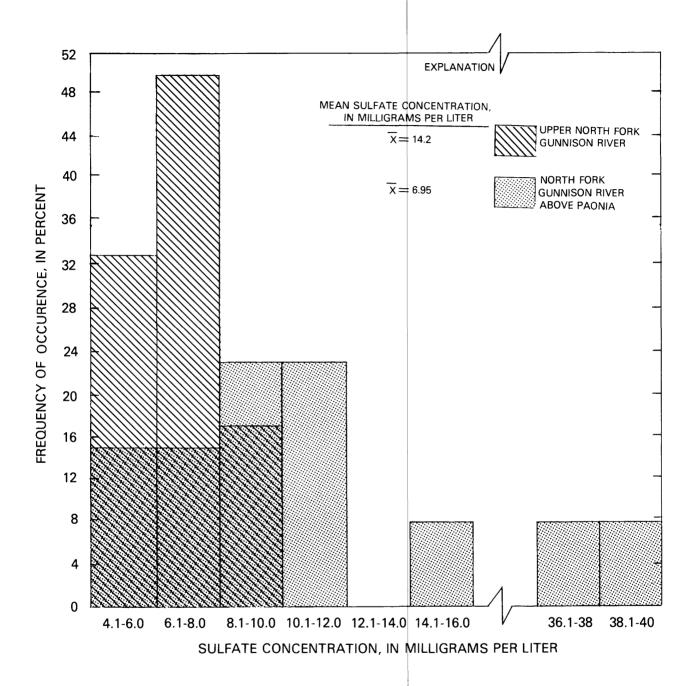


Figure 12.--Frequency of occurrence of sulfate concentrations for sites on the North Fork Gunnison River, upstream and downstream from coal-mining activities.

#### SUMMARY AND CONCLUSIONS

Following evaluation of differences in geology and elevation, and statistical analyses of the water-quality data, the study area was divided into four subareas for data analyses. Location of the subareas included major watershed divides in the study area. These subareas were the Muddy Creek, Anthracite Creek, Hubbard and Terror Creeks, and the North Fork Gunnison River.

Dissolved-solids concentration and specific-conductance values from the study area were small. Dissolved-solids concentrations in the area ranged from 34~mg/L to 337~mg/L with the smallest mean concentration in the Anthracite Creek subarea (55~mg/L) and the largest mean concentration in the Hubbard and Terror Creeks subarea (133~mg/L). Mean dissolved-solids concentration for the study area was 97~mg/L.

Dissolved constituents generally followed the same pattern as dissolvedsolids concentration and were, for the most part, small. For most dissolved constituents, mean concentrations were smallest in the Anthracite Creek subarea, and largest in the Hubbard and Terror Creeks subarea.

Suspended-sediment concentrations in the study area ranged from less than 1 mg/L to 5,790 mg/L, with the smallest mean concentration in the Anthracite Creek subarea (122 mg/L), and the largest mean concentration in the Muddy Creek subarea (361 mg/L). Mean suspended-sediment concentration for the entire study area was 214 mg/L. Mean suspended-sediment discharge for the study area was 355 ton/d, with the largest mean suspended-sediment discharge in the North Fork Gunnison River subarea (693 ton/d).

Of the trace elements analyzed, mean total concentrations were largest in the Muddy Creek subarea, and smallest in the Hubbard and Terror Creeks subarea. The study area appeared to have relatively large total-iron concentrations compared to other measured trace elements; the cause of this large concentration is unknown, but it probably is related to local geology.

Paonia Reservoir on Muddy Creek appeared to have substantial effects on suspended-sediment and trace-element concentrations. Mean suspended-sediment concentration decreased by nearly 6.5 times (from 862 mg/L to 132 mg/L) between inflow and outflow of the reservoir and mean total trace-element concentrations decreased by an average of nearly 67 percent. The reservoir appeared to have only a minor effect on major-dissolved constituents, although the concentrations of all dissolved constituents decreased slightly from upstream to downstream from the reservoir.

The amount of data available is inadequate to determine what effects coal mining in the area may have on the water quality. Analyses of the data showed slight increases in alkalinity, sulfate, and dissolved-solids concentrations, but it is unknown if these increases were caused by coal-mining activities or downstream effects such as evaporation and ground-water inflow. More data are needed to determine what effects coal-mining activities in the area may have on the water quality.

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